



## SCRUTINY OF GEOPOLYMER-LATERITIC SOIL USING CLUSTER AND PIPER PLOT STATISTICAL AIDS

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### ABSTRACT

*The strength of a fine-grained lateritic soil from three (3) different localities on Abuja – Lokoja road where road failure happen was treated with rice husk ash (RSA), cement and sodium silicate activator (SSA), with varying percentage examined by means of Atterberg, Compaction and triaxial shear tests. The result showed that Conductivity has the highest mean (144.67 $\mu$ S/cm) and followed by TDS (94.00mg/l) while salinity recorded the lowest (0.057mg/l). Also, TDS recorded the highest Standard deviation (1.10mg/l). This was followed by TSS (0.75mg/l) while salinity recorded the least value of (0.0006mg/l). Further, all the parameter of the water samples examined are within the maximum permissible limit of construction works standards. Also correlation analysis for CBR and UCS shows a positive correlation (between the CBR, RHA and the OMC which indicate a direct relationship between the variables. A significant correlation was found to exist between Geopolymer and Kaolin clay powder ( $r = 0.95, \alpha = 0.05$ ).*

**KEYWORDS:** Construction material, Sodium silicate, Geopolymer, Kaolin clay, Abuja.

### INTRODUCTION

Buildings made of earth materials are the most common inexpensive accommodation since earth materials are readily accessible almost anywhere on earth (Pooria et al. 2021; Xu et al. 2021; Zhu et al. 2020). It has been initiate that lateritic soils are in general good construction materials, and for that reason are commonly utilized in construction (Adbullahi 2021; Watez 2021; Rivera et al. 2020). Chemical improvement or stabilization is the reform of properties of a locally accessible soil, in order to improve its engineering performance. The two most frequently utilized chemical stabilization techniques are cement and lime stabilization (Venkatesh 2021; Adeyanju et al. 2020; Wang et al. 2020). On the other hand, owing to the Carbon Dioxide (CO<sub>2</sub>) emissions generated during the manufacturing of aforesaid soil improvers or stabilizers, which contribute to greenhouse gases, other soil enhancers are being sought out and suggested (Zhu et al. 2020; Abdullah et al. 2020). An unconventional to stabilizing soil is via introducing geopolymer materials as well as activators. The multivariate statistical approach is a technique which deals with data that consist of of sets of quantities or a number of individuals or objects variables as declared by Achieng et al., (2017). Multivariate approach can aid simplification and cataloguing of big data sets, and also suitable for drawing significant suppositions as observed by Bodrud-Dova et al. (2016). Amongst the multivariate statistical approaches, the hierarchical cluster scrutiny or analysis (HCA) and principal components scrutiny or analysis (PCA) are mostly used in feature studies. In cluster scrutiny or analysis (CA), the data cluster is distributed into groups in terms of the connexions and dissimilarities.

Pearson connexion coefficient measures the linear correlation among X and Y. Based on Cauchy-Schwarz disparity, it has a value ranges from +1 to -1, where 1 is total positive linear connexion (Ahmed et al. 2018; Achieng et al. 2017). Meanwhile, a piper plot comprised of three sections: a ternary illustration in the lower left signifying cations (sodium plus potassium, magnesium, and calcium), a ternary illustration in the lower right signifying anions (carbonate plus bicarbonate, chloride, and sulphate), and a diamond map in the



middle which is a matrix alteration of the two ternary illustrations. Each sample is standardised to one hundred (100) that is sum of cations equal to one hundred, and sum of anions equal to one hundred), so that the comparative concentrations are on a percentage basis. Furthermore, the durov'splotvalidates that there are collaboration means of two or more differentfacies that occurred in soil surface system (Ogwueleka and Igibah2020; Igibah and Apagu2019).

## 2. LITERATURE REVIEW

### 2.1 Concept of Rice Husk

Rice with botanica name of *Oryza sativa* L, is a main source of food for billions of populace and among the major crops on earth (Dheyab et al. 2019; Alshaba et al. 2018; Wen et al. 2019). It covers around one percent (1%) of the earth's surface. Rice husk or fibre(RH) is an economical byproduct of rice processing that was separated from rice grain during the rice refining process. It is testified that, for every one ton of rice manufactured, roughly 0.23 tons of RH is produced. In the paddy foliage ofNigeria with a land region of approximatelyninety-one (91) millionshectares for arable and manufactured a total of three million (3,000,000)tonnes of rice grains yearly (Rahgozar and Saberian2018; Roychand2021; Rivera 2020).

The quantity of husksproduced is twenty percent (20%) of three million (3,000,000)tonnes which is six hundred thousand (600,000)tonnes.The key components and physical features of RH are tabulated in Table 1.

Despite the factthat some of this husk is transformed into end products such as panel boards and activated carbon, supplementary cementing material and insulating material, however, like many other agricultural byproducts, the industrial usages of this biomass are still limited with little economic value (Seyhan et al. 2020; Saberian e al. 2020; Khasib and Daud2020). Hence, it is very vital to find pathways to fully utilize the RH and an intense investigation scrutiny is presently undertaken globally to ascertain potential applications as well as develop economically feasible methods for these usages on a commercial scale (Sharma et al. 2019; Tan et al. 2019; Teing et al. 2019).

Table 1: Properties of rice

| Properties                | Rice Husk | Physical properties   |
|---------------------------|-----------|---|
| Components (%)            |           |   |
| Cellulose                 | 25-35     | Particle size ( $\mu\text{m}$ ) – 26.64<br>Surface area ( $\text{m}^2/\text{g}$ ) – 0.92<br>Density ( $\text{g}/\text{cm}^3$ ) – 1.00 |
| Hemicellulose             | 18-21     |   |
| Lignin                    | 26-31     |   |
| Silica ( $\text{SiO}_2$ ) | 15-17     |   |
| Solubles                  | 2-5       |   |
| Moisture content          | 5-10      |   |

### 2.2 Geopolymers obtained Kaolin clay

Kaolin, also named china clay is a soft white clay that is a vitalfeature in the manufacturing of paper, rubber, paint, porcelain, and many other products (Amiri and Emami2019; Chang and Cho 2019).

In its natural state be made of principally the mineral kaolinite, which, beneath the electron microscope, is seen to compriseapproximately hexagonal, platy crystals ranging between point one (0.1)micrometre to ten (10)micrometres or even larger as illustrated in Figure 1.



**Figure 1: Rice husk and Kaolin clay.**

### **3.0 ANALYSIS RESULTS**

#### **3.1 Descriptive statistics**

Figure 2a & b. and Table 2, describes the physicochemical parameters of water used in analysis cement, RHA, KCP and geopolymer mixtures on the Atterberg limit, compaction, CBR and UCS characteristics of the soils tested. The result showed that Conductivity has the highest mean (144.67 $\mu$ S/cm) and followed by TDS (94.00mg/l) while salinity recorded the lowest (0.057mg/l). Also, TDS recorded the highest Standard deviation (1.10mg/l). This was followed by TSS (0.75mg/l) while salinity recorded the least value of (0.0006mg/l). Further, all the parameter of the water samples examined are within the maximum permissible limit of construction works standards.

#### **3.2 Cluster Analysis (Using Squared Euclidean distance)**

Result from Figures 3 detect similarity groups between the lateritic soils and geopolymer. Two statistically significant clusters are formed for Atterberg C1 and C2, which yielded two groups of similarity between the sampling sites. CBR, compaction, UCS and Triaxial test represented are presented as A and B. Temporal cluster analysis was used on standardized log-transformed data sorted by season. CA was performed using squared Euclidean distances as a measure of similarity (Agashua and Ogiye 2018; Abdullahi et al. 2020; Igibah et al. 2020).

#### **3.3 Impact of Correlation Analysis (Atterberg limit, compaction, CBR and UCS)**

The laboratory experimental results analysis for this research work was carried out using Correlation analysis as the primary statistical tool and statistical package for social science (SPSS) as statistical software packages for statistical analysis. Correlation analysis is a statistical technique to quantify the dependence of two or more variables (Ahmed et al. 2018; Amiri and Emami 2019). The purpose of a correlation analysis is to determine whether there is a relationship between sets of variables CBR, RHA, and OMC or UCS, RHA, and OMC and the Outcomes of correlation are summarized in Table 3. Inference was based on the strength tests (CBR soaked and UCS) results gotten from cement, Rice Husk Ash (RHA), kaolin clay powder and geopolymer mix stabilized soil other than those obtained from RH. Each of the parameters pair were calculated using Pearson's correlation coefficients as displayed in Table 4. A significant correlation was found to exist between Geopolymer and Kaolin clay powder ( $r = 0.95$ ,  $\alpha = 0.05$ ).

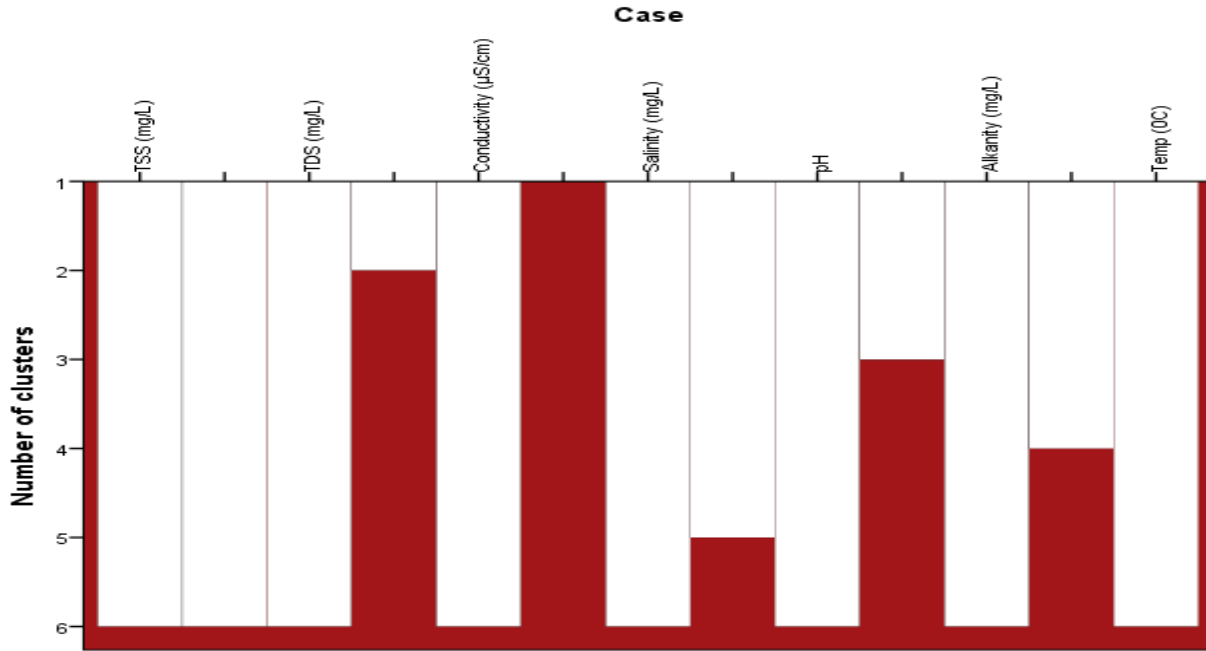


Figure 3: Numbers of clusters for water test

Table 3: Relationship between UCS, cement, RHA, KCP and OMC (H<sub>1</sub>: there is significant relationship between, CBR, OMC and geopolymers).

| Variable   | N | Mean   | Std Dev | r-cal | P-value | Remark                |
|------------|---|--------|---------|-------|---------|-----------------------|
| Stabilizer | 6 | 441.97 | 191.88  |       |         |                       |
| Cement     | 6 | 64.06  | 21.63   |       |         |                       |
| RHA        | 6 | 74.07  | 25.21   | 0.980 | 0.003   | Reject H <sub>0</sub> |
| KCP        | 6 | 358.20 | 145.04  |       |         |                       |

### 3.4 Water quality assessment (Piper, Ternary, Durov and Schoeller plots)

Figures 4-7 the polar class and wind, piper diagram ternary, surfer and contour, shows that the Stations 1, 2 and 5 are dominant in sodium chloride type of water. The station 3 is dominant in the Calcium–chloride type of water, while station 4 is dominant in mixed type of water, which means no cations and anions exceeds 50%. Also, the results revealed that 43.7% of the samples could be classified as S04-Cl-Ca - Mg type, 20.2% of the samples as S04 - Cl-CO<sub>3</sub> HC0<sub>3</sub>-Ca-Mg type and 37.1% of the samples as Mg - Ca type. Studies including Achieng et al. (2017) and Ogwueleka and Igibah (2019) also found comparable results by identified major water types and their ionic composition of different water samples. The calcium sulphate and calcium–chloride are rich in these regions because the parent rock are of Fissile Hornblende Biotite gneiss and Charnockite, which have composed of sodium and calcium rich minerals. (Igibah and Tanko 2019; Wen et al. 2019).

Durov’s sand scatter diagram aids in the clarification of the evolutionary styles and the hydrochemical processes happening in the surface water system and can show mixing of various water types, converse ion exchange and ion exchange. The results from Figures 8 & 9 reveals water in the box 1 as carbonated group which symbolizes water rich in Ca-HCO<sub>3</sub>, HCO<sub>3</sub>-CO<sub>3</sub> and TDS thus revitalizing. The water from the center of the plain is positioned in triangle 1 and 2, rich in Ca-Cl plus Mg-SO<sub>4</sub> which triggering ion exchange and dissolution of vaporize minerals.



## CONCLUSION

The statistical analysis for CBR and UCS shows a positive correlation (between the CBR, RHA and the OMC which indicate a direct relationship between the variables. A significant correlation was found to exist between Geopolymer and Kaolin clay powder ( $r = 0.95$ ,  $\alpha = 0.05$ ). Piper plot diagram shows the Stations 1, 2 and 5 are dominant in sodium chloride type of water. The station 3 is dominant in the Calcium–chloride type of water, while station 4 is dominant in mixed type of water, which means no cations and anions exceeds 50%. Also, the results revealed that 43.7% of the samples could be classified as S04-Cl-Ca - Mg type, 20.2% of the samples as S04 - Cl- CO<sub>3</sub> HCO<sub>3</sub>-Ca-Mg type and 37.1% of the samples as Mg - Ca type which is good for construction purposes. The Durov's diagram demonstrates that there are collaboration processes of two or more various facies that happened in water system. The water from the center of the plain, is positioned in triangle 1 and 2, rich in Ca-Cl plus Mg-SO<sub>4</sub> which triggering ion exchange and a dissolution of vaporize minerals and make the water suitable for road works. Based on the EDAX as well as SEM outcomes the new produced gel-like cementitious compounds of sodium aluminosilicate hydrate (N–A–S–H) were assumed to be the major cause of strength improvement. The XRD results also revealed a general increase in the lateritic soil mineral peaks.

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**APPENDIX**

**Table 2: Descriptive analysis for water test**

|                        |         | Temp (0C) | pH     | Conductivity (µS/cm) | TDS (mg/L) | TSS (mg/L) | Alkanity (mg/L) | Salinity (mg/L) |
|------------------------|---------|-----------|--------|----------------------|------------|------------|-----------------|-----------------|
| N                      | Valid   | 6         | 6      | 6                    | 6          | 6          | 6               | 6               |
|                        | Missing | 1         | 1      | 1                    | 1          | 1          | 1               | 1               |
| Mean                   |         | 26.7833   | 6.3667 | 144.6667             | 94.0000    | 91.8333    | 53.0000         | .0570           |
| Std. Deviation         |         | .11690    | .05164 | .51640               | 1.09545    | .75277     | .63246          | .00063          |
| Variance               |         | .014      | .003   | .267                 | 1.200      | .567       | .400            | .000            |
| Skewness               |         | -.668     | -.968  | -.968                | -1.369     | .313       | .000            | .000            |
| Std. Error of Skewness |         | .845      | .845   | .845                 | .845       | .845       | .845            | .845            |
| Kurtosis               |         | -.446     | -1.875 | -1.875               | 2.500      | -.104      | 2.500           | 2.500           |
| Std. Error of Kurtosis |         | 1.741     | 1.741  | 1.741                | 1.741      | 1.741      | 1.741           | 1.741           |
| Range                  |         | .30       | .10    | 1.00                 | 3.00       | 2.00       | 2.00            | .00             |
| Minimum                |         | 26.60     | 6.30   | 144.00               | 92.00      | 91.00      | 52.00           | .06             |
| Maximum                |         | 26.90     | 6.40   | 145.00               | 95.00      | 93.00      | 54.00           | .06             |

**Table 4: Correlation of water samples used for geopolymer stabilization**

|                      |                     | Temp (0C) | pH    | Conductivity (µS/cm) | TDS (mg/L) | TSS (mg/L) | Alkanity (mg/L) | Salinity (mg/L) |
|----------------------|---------------------|-----------|-------|----------------------|------------|------------|-----------------|-----------------|
| Temp (0C)            | Pearson Correlation | 1         | .883  | .221                 | .000       | .417       | .271            | .541            |
|                      | Sig. (2-tailed)     |           | .020  | .674                 | 1.000      | .411       | .604            | .268            |
| pH                   | Pearson Correlation | .883      | 1     | .250                 | .354       | .343       | .000            | .612            |
|                      | Sig. (2-tailed)     | .020      |       | .633                 | .492       | .506       | 1.000           | .196            |
| Conductivity (µS/cm) | Pearson Correlation | .221      | .250  | 1                    | .354       | .857       | -.612           | .612            |
|                      | Sig. (2-tailed)     | .674      | .633  |                      | .492       | .029       | .196            | .196            |
| TDS (mg/L)           | Pearson Correlation | .000      | .354  | .354                 | 1          | .243       | -.866           | .577            |
|                      | Sig. (2-tailed)     | 1.000     | .492  | .492                 |            | .643       | .026            | .230            |
| TSS (mg/L)           | Pearson Correlation | .417      | .343  | .857                 | .243       | 1          | -.420           | .840            |
|                      | Sig. (2-tailed)     | .411      | .506  | .029                 | .643       |            | .407            | .036            |
| Alkanity (mg/L)      | Pearson Correlation | .271      | .000  | -.612                | -.866      | -.420      | 1               | -.500           |
|                      | Sig. (2-tailed)     | .604      | 1.000 | .196                 | .026       | .407       |                 | .312            |
| Salinity (mg/L)      | Pearson Correlation | .541      | .612  | .612                 | .577       | .840       | -.500           | 1               |
|                      | Sig. (2-tailed)     | .268      | .196  | .196                 | .230       | .036       | .312            |                 |

a. Listwise N=6

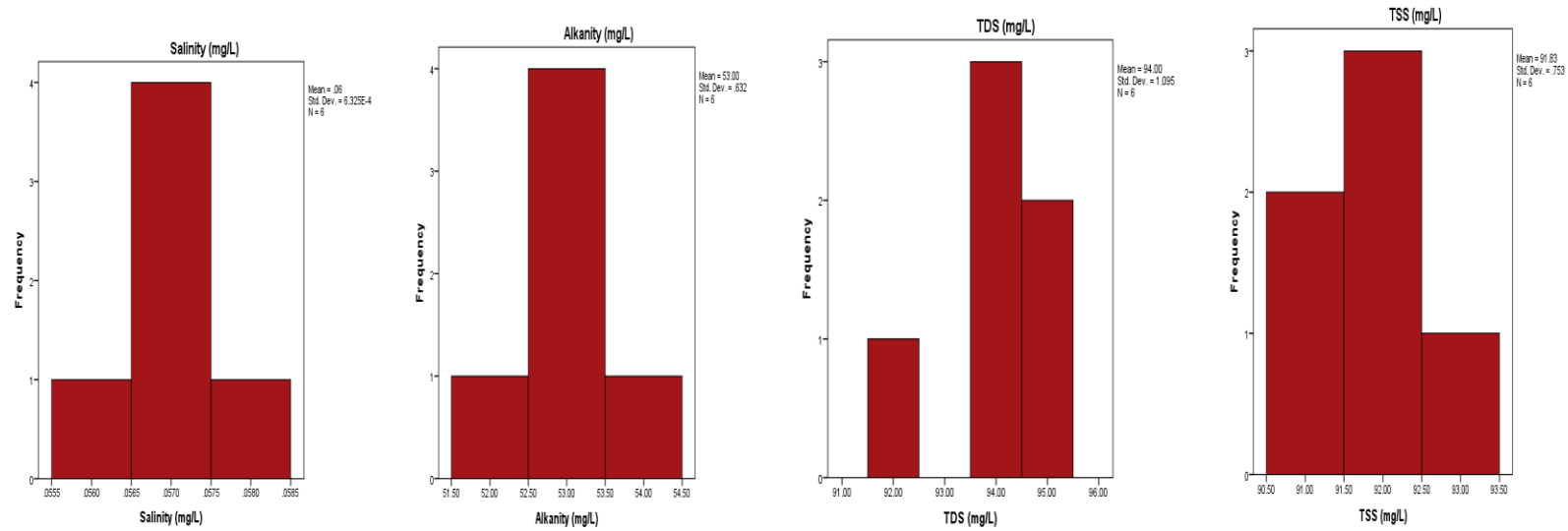


Figure 2a: Histogram showing Salinity, Alkalinity, TDS and TSS of water utilized for analysis.



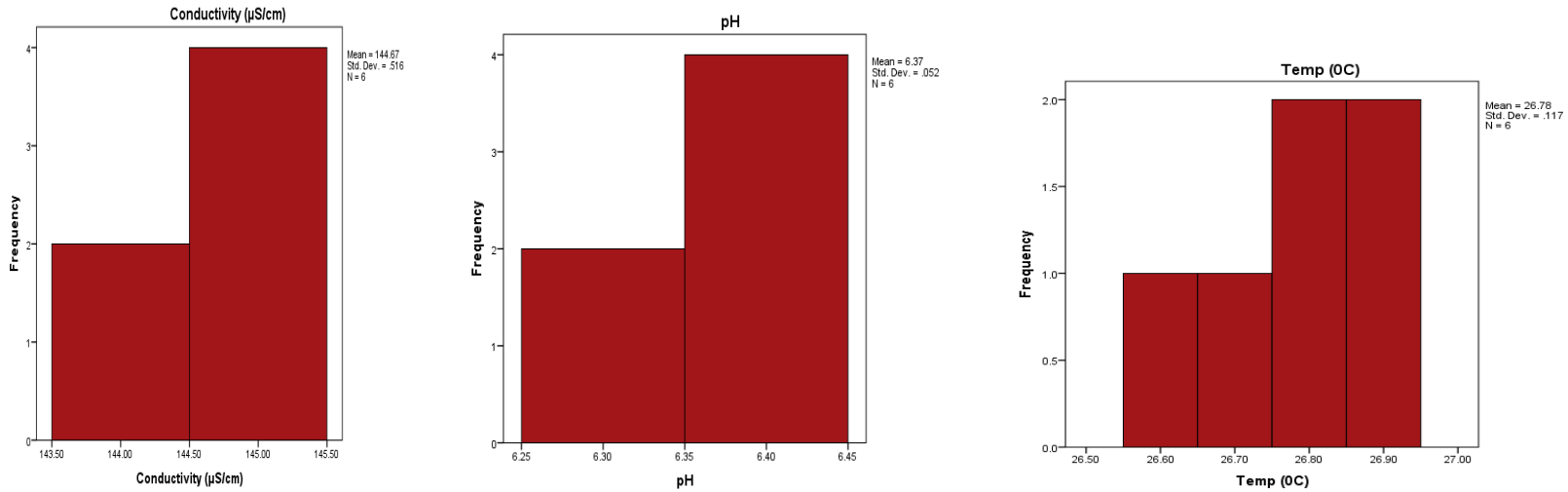


Figure 2b: Histogram showing Conductivity, pH and Temperature of water utilized for analysis.

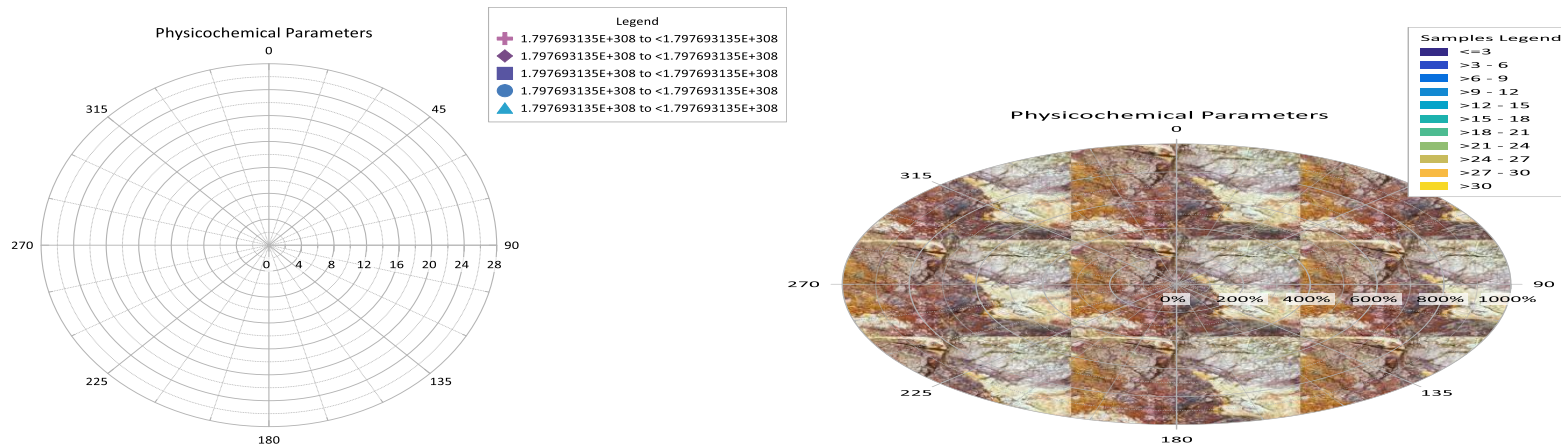


Figure 4: Polar class and wind plot of water analysis.

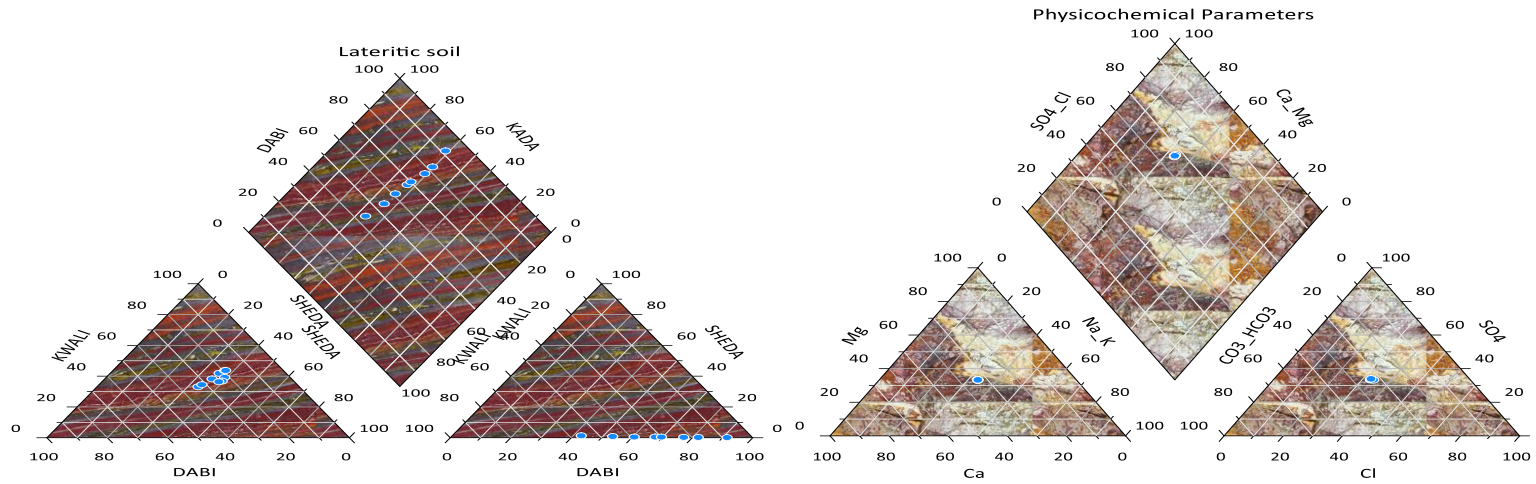


Figure 5: Piper diagram displaying the hydrological looks of the (a) lateritic soil, (b) water samples.

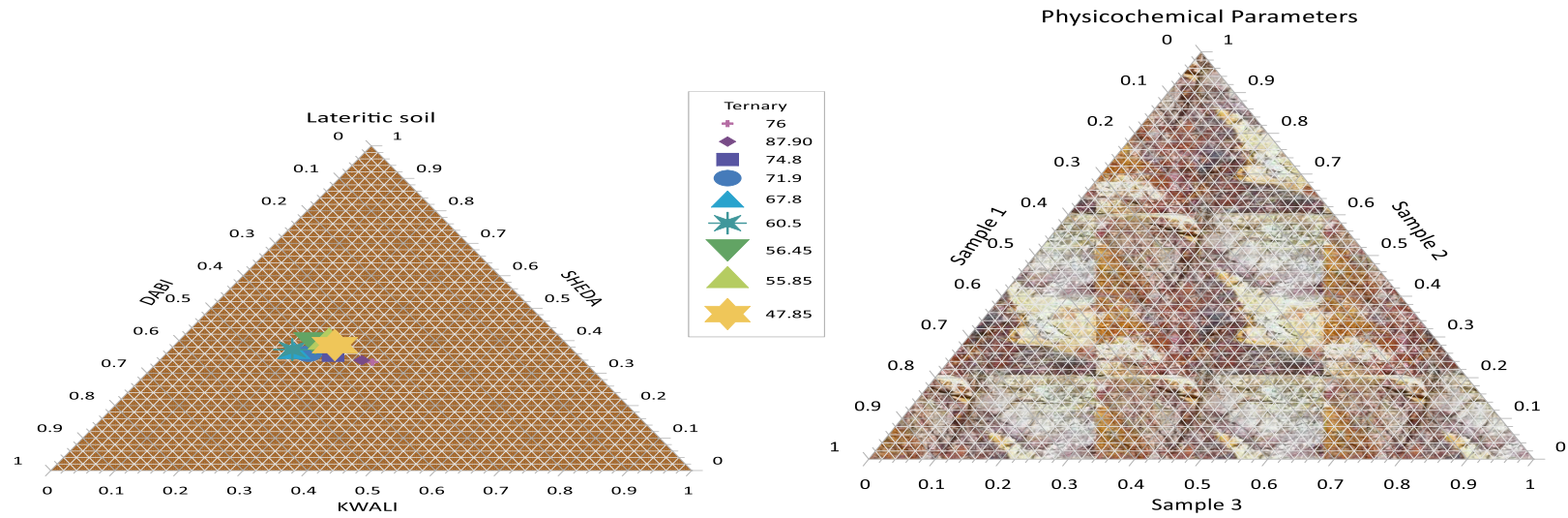


Figure 6: Ternary diagram displaying the hydrological looks of the (a) lateritic samples, (b) water samples.

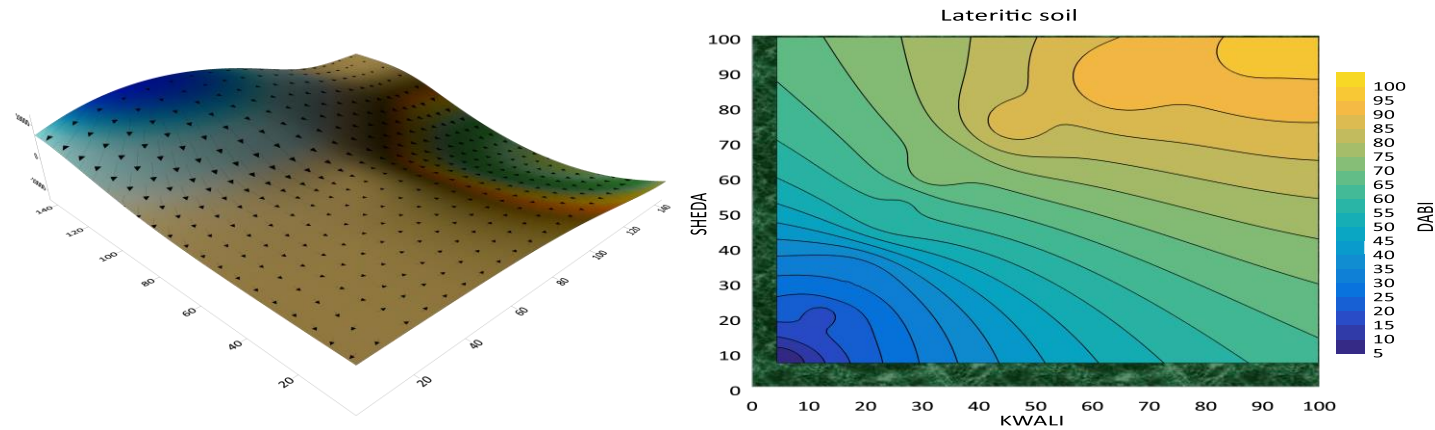


Figure 7: Surfer and contour of lateritic soil

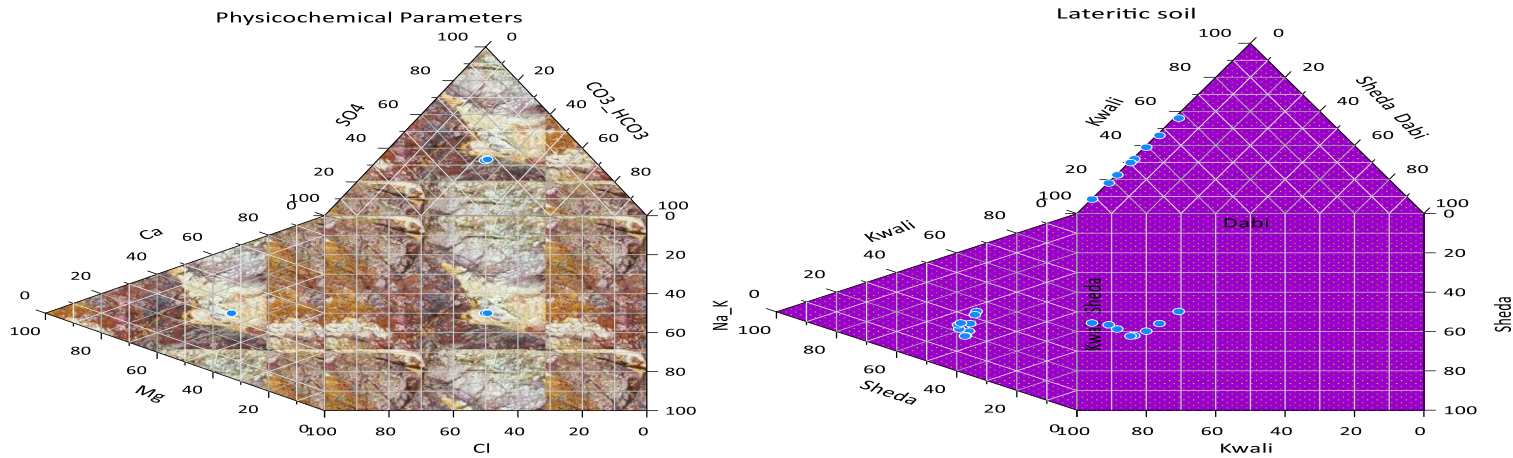


Figure 8: Durov displaying the hydrological looks of the: (a) water samples, (b) lateritic samples.

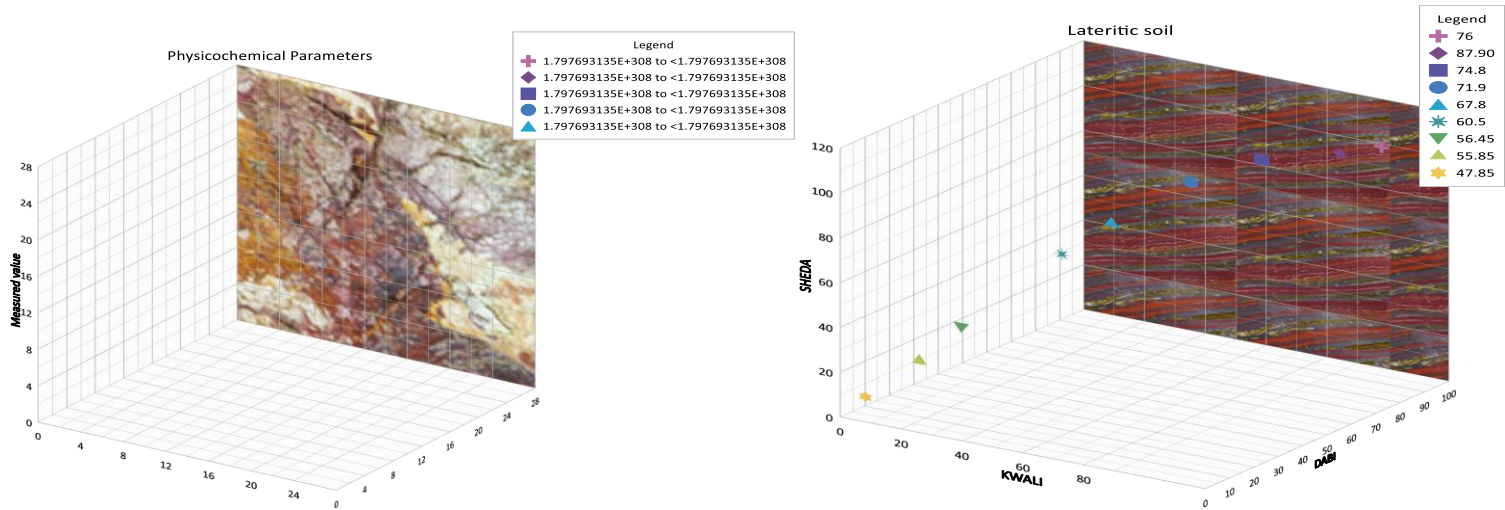


Figure 9: scatter diagram displaying the hydrological looks of the: (a) water samples, (b) lateritic soil.